Wavefront Velocity Oscillations of Carbon-Nanotube-Guided Thermopower Waves: Nanoscale Alternating Current Sources

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here is a growing interest in the development of micro- or nanosized energy storage^{1,2} and harvesting systems³⁻⁹ that are capable of releasing energy at high rates. This is motivated by emerging applications such as so-called "smart dust",¹⁰ microelectromechanical systems,¹¹ disposable medical diagnostics,^{12,13} and flexible electronic devices.¹⁴ However, power sources capable of the direct conversion of chemical energy to oscillating current do not exist, despite obvious applications of alternating current waveforms such as cardiac defibrillation and pacemaker signals,^{12,15} micropiezoelectric oscillators,⁴ RF identification tags,¹⁶ and micromixers.¹⁷ In this work, we investigate the nonlinear coupling between exothermic chemical reactions and a nanotube or nanowire with large axial heat conduction, resulting in selfpropagating thermal waves guided along the nanoconduit. The resulting reaction wave induces a concomitant thermopower wave of high power density (>7 kW/kg), resulting in an electrical current that travels along the same direction. We show that such waves are predicted to demonstrate oscillatory front velocities under certain values of the chemical reaction rate constants and thermal parameters. These oscillations are verified experimentally using a cyclotrimethylene-trinitramine (TNA)/multiwalled carbon nanotube (MWNT) system and find a dominant frequency at approximately 430 Hz. We show that the oscillations and the frequency dispersion are well-described using a diffusive heat transfer model with a nonlinear source term coupled to a reactive mass balance along the nanoconduit.

ABSTRACT The nonlinear coupling between exothermic chemical reactions and a nanowire or nanotube with large axial heat conduction results in a self-propagating thermal wave guided along the nanoconduit. The resulting reaction wave induces a concomitant thermopower wave of high power density (>7 kW/kg), resulting in an electrical current along the same direction. We develop the theory of such waves and analyze them experimentally, showing that for certain values of the chemical reaction kinetics and thermal parameters, oscillating wavefront velocities are possible. We demonstrate such oscillations experimentally using a cyclotrimethylene-trinitramine/multiwalled carbon nanotube system, which produces frequencies in the range of 400 to 5000 Hz. The propagation velocity oscillations and the frequency dispersion are well-described by Fourier's law with an Arrhenius source term accounting for reaction and a linear heat exchange with the nanotube scaffold. The frequencies are in agreement with oscillations in the voltage generated by the reaction. These thermopower oscillations may enable new types of nanoscale power and signal processing sources.

KEYWORDS: carbon nanotubes · cyclotrimethylene-trinitramine · thermopower · power source · oscillation · reaction diffusion · energy storage

Carbon nanotubes (CNTs) have several advantages for propagating these thermopower waves, despite possessing unexceptional Seebeck coefficients (approximately 80 μ V/K).¹⁸ They remain stable during high-temperature anaerobic reactions, and their axial thermal conductivity is exceedingly high,¹⁹ resulting in large reaction front velocities and therefore large specific power outputs. The surface-to-volume ratios are also large for ordered 2D and 3D arrays of nanotubes, resulting in sizable heat exchange rates. Conversely, high Seebeck coefficient materials such as Bi₂Te₃ and Sb₂Te₃ are counterintuitively not favorable for guiding thermopower waves due to their low thermal conductivities.

In this work, we use cyclotrimethylenetrinitramine (TNA) as the fuel source due to its large enthalpy of reaction and strong adhesion to CNTs. CNTs are wrapped in TNA, *Address correspondence to strano@mit.edu.

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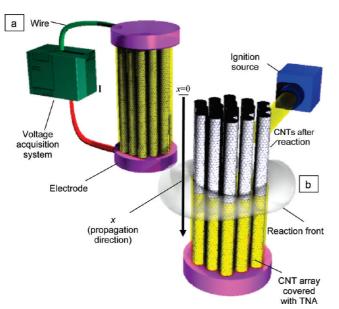


Figure 1. (a) Schematic of the thermopower wave generator. (b) Ignition at one end of a cyclotrimethylene-trinitramine/multiwalled carbon nanotube (TNA–CNT) array results in an exothermic decomposition reaction that propagates along the CNT *via* its high axial thermal conductivity. The resulting thermal wave develops a concomitant thermopower wave of high power density.

as shown in Figure 1; the reaction occurs at one end of this TNA-CNT system and propagates with the help of CNT conduits. We have previously demonstrated large power densities using TNA-CNT systems.²⁰ In this work, we show that the theory of such selfpropagating thermal waves predicts distinct regimes where oscillatory wavefront velocities are possible and demonstrate their existence experimentally. Both theory and measurements are in agreement, with oscillations ranging in frequency from 400 to 5000 Hz.

Theory of Thermopower Wave Oscillations. We consider an annulus of material capable of first-order, exothermic decomposition (such as TNA) surrounding a carbon nanotube. Initiation of the chemical decomposition releases heat that travels much faster along the nanotube than in the annulus, resulting in a reaction wave guided by the CNT. We consider the diffusive transport limit where the reaction zone length exceeds that of the phonon mean free path in the CNT. The coupled thermal diffusion and chemical reaction terms can then be described using a modified Fourier's Law equation with a nonlinear source term to account for the chemical reaction. Under adiabatic conditions, the equation for the temperature of the annulus (*T*) becomes

$$\frac{\rho C_{\rm p}}{M_{\rm W}} \frac{\partial T}{\partial t} = \chi \frac{\partial^2 T}{\partial x^2} - (Qk_0 y)e^{(-E_0/RT)} - G_0 \frac{4d_{\rm N}}{d_{\rm M}^2 - d_{\rm N}^2} (T - T_2)$$
(1)

where G_0 is the interfacial conductance between the CNT and TNA, d_M is the outer diameter of the TNA–CNT system, and d_N is the CNT diameter. The parameter χ is the thermal conductivity, ρ is the density, Q is the en-

thalpy of reaction, *y* is the molar concentration of reactive material, k_0 is the Arrhenius prefactor, *R* is the universal gas constant, E_a is the activation energy of decomposition, and C_p and M_W are the (molar) specific heat and molecular weight of the material, respectively. The second term on the right side accounts for the thermal source of the decomposition reaction, with the mole fraction of unreacted material (y/y_0) decreasing in time and space (according to the spatial profile of *T*) according to

$$\frac{\partial y}{\partial t} = -(k_0 y)e^{(-E_a/RT)}$$
(2)

The corresponding equation for the temperature (T_2) evolution in the carbon nanotube includes only terms for the heat transfer from the annulus and thermal diffusion:

$$\frac{\rho_2 C_{p,2}}{M_{W2}} \frac{\partial T_2}{\partial t} = \chi_2 \frac{\partial^2 T_2}{\partial x^2} + G_0 \frac{4}{d_N} (T - T_2)$$
(3)

where the subscript "2" refers to properties of the CNT. Equations 1 through 3 are solved using the following boundary conditions.

$$T_1(x, t = 0) = g \times e^{-x^2/w} + T_{\text{ambient}}$$
$$T_2(x, t = 0) = g \times e^{-x^2/w} + T_{\text{ambient}}$$
$$\frac{y}{y_0}(x > 0, t = 0) = 1$$

 $\frac{y}{y_0}(x=0,t) =$

0.001 (to approximate a zero boundary condition)

with the pulse width parameter $w = 2 \times 10^{-11} \text{ m}^2 \cdot \beta$. T_{ambient} is the temperature of the surroundings, assumed to be 298 K. The parameter grouping $\beta = (C_p E_a)/2$ (-QR) is the inverse dimensionless adiabatic reaction temperature rise, which is shown to be important for determining wave characteristics. The function T(x, t =0) is a Gaussian pulse, offset by 298 K, to simulate initiation by a rapid pulse of heat, for example, from a laser or resistive filament. Here, g is the maximum initial temperature. From numerous simulations, we determined empirically that it must be greater than 0.16 imes $exp(0.37\beta)$ to initiate a propagating reaction wave. Physically, the magnitude of the temperature initial condition must increase with β since that parameter is proportional to specific heat and activation energy. More energy must be supplied in the initial pulse for larger values of β to overcome specific heat and activation energy and increase the initial reaction rate sufficiently to sustain self-propagation. The initial condition for the composition of the material, y/y_0 , corresponds to a completely unreacted annulus, with the exception of the x = 0 boundary, which is almost entirely reacted so as to be consistent with the temperature condition, as

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is necessary for numerical stability. The boundaries of the system are assumed to be adiabatic, and the system is made to be sufficiently long such that the wave is far from the boundaries. Further details of the simulation are given in the Methods section and in Supporting Information.

Our work is the first to examine this mathematical system of two coupled heat transport equations in conjunction with a chemical reaction. We have studied variants of these equations previously for the case of nanotube-guided nanothermites.²¹ We also note that simplified versions of eqs 1 and 2 have been studied historically to describe 1D combustion waves and combustion synthesis.²²⁻²⁵ Following Zel'dovich and Frank-Kamenetskii,²⁶ we nondimensionalize the system of equations. A nondimensional temperature u = (R/ E_a)T, time $\tau = (-Qk_0R)/(C_pE_a)t$, and space $\xi = x((\rho C_p/2))$ $\chi M_{\rm W}$) $(-Qk_0R/C_{\rm p}E_{\rm a}))^{1/2}$ are shown to appropriately scale the system and resulting reaction wave.²³ A conversion η can be defined ($M_W y$)/ $\rho = 1 - \eta$. Other parameters of importance are the thermal diffusivities $\alpha = (\chi M_W / \rho C_p)$ of the annulus and of the nanotube, α_2 . The resulting equations are

$$\frac{\partial u}{\partial \tau} = \frac{\partial^2 u}{\partial \xi^2} + (1 - \eta)e^{-1/u} - \gamma_1(u - u_2)$$
(4)

$$\frac{\partial u_2}{\partial \tau} = \alpha_0 \frac{\partial^2 u_2}{\partial \xi^2} + \gamma_2 (u - u_2) \tag{5}$$

$$\frac{\partial \eta}{\partial \tau} = \beta (1 - \eta) e^{-1/u} \tag{6}$$

where $\alpha_0 = \alpha_2 / \alpha_r \gamma_1 = (4 d_N / d_M^2 - d_N^2) (G_0 \beta M_W / \rho C_p k_0)$ and $\gamma_2 = (4G_0\beta M_{W2})/(d_N\rho_2 C_{p,2}k_0)$. Interestingly, this shows that the only terms affected by the diameter of the thermal conduit or the fuel layer are the interfacial heat exchange terms, γ_1 and γ_2 . We have demonstrated with previous modeling work^{20,21} that γ_1 and γ_2 do not affect the propagation of the reaction wave for values above 10⁻³; axial thermal transport within the nanotube becomes the limiting factor compared to interfacial conductance. By comparison, the simulations in this article used values on the order of 10³, so diameters have little effect to first order. A central motivation for this work is the fact that oscillatory behavior has been observed in similarly driven Fourier conduction systems;^{23,27-30} hence, we ask if thermopower waves can likewise demonstrate regions of parameter space where oscillations are possible.

For β , α , and α_2 , average values over the temperature range of 300 to 1700 K were used. Since β is the nondimensionalized adiabatic reaction temperature rise, the maximum temperature was iterated until a consistent average β of 10.6 for TNA was reached. Over the same range, α is 4.3 \times 10⁻⁸ m²/s and α_2 is 0.001 m²/s. Depending on the concentration of defects, the thermal diffusivity of the nanotube, α_2 , may at least vary

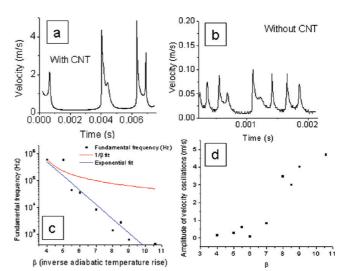


Figure 2. (a) Simulation of the wavefront velocity using $\beta = 10.6$, $\alpha = 4.3 \times 10^{-8} \text{ m}^2/\text{s}$, and $\alpha_2 = 0.001 \text{ m}^2/\text{s}$, with a length of 5 mm. The diameter of the nanotube was 22 nm, and the surrounding fuel annulus was 7 nm thick. (b) Simulated velocity for the same parameters with no nanotube ($\alpha_2 = 0$). (c) Fundamental frequency variation with β . The relation is inverse and stronger than simple dimensional scaling (1/ β) would suggest. (d) Amplitude of oscillations for different β values. Amplitudes are small and relatively constant for $\beta < 7$.

over an order of magnitude. The value chosen here was the median. For all other parameters except k_0 , the same values were used as in previous studies.²⁰

The parameter k_0 affects the scaling of both the reaction wavefront velocity (henceforth referred to as velocity) and oscillation frequencies when they are converted from the nondimensional solutions of eqs 4-6 through the space and time scaling factors. The velocity and frequency depend on $k_0^{1/2}$ and k_0 , respectively. Since k_0 is an *effective* Arrhenius constant, it can include mass and thermal diffusion limitations. We argue that the presence of the nanoconduit may reduce k_0 for the reacting material, TNA. While the MWNTs act as thermal conduits, kinetically they are inert, so their effect on k_0 may be more similar to that of a binder. Oyumi³¹ measured k_0 in pure TNA to be 3.75 \times 10¹⁸ s⁻¹, whereas Williams and Matei³² found it to be 2.0 imes 10¹¹ s⁻¹ in a TNA/binder mixture over the range of 180 to 225 °C. Consequently, in order to reproduce the reaction wave velocity in the range of 0.1 to 2 m/s, which was in agreement with the experiments, $k_0 = 3 \times 10^9 \text{ s}^{-1}$ was chosen. The validity of the choice was further confirmed by the excellent agreement between the simulation and measurements signature frequencies, which will be described later.

We first note that the simulations predict velocity oscillations for the TNA–CNT system. Figure 2a shows the velocity profile, beginning at t = 0.2 ms in order to see the steady-state responses after the initial period, which is more strongly influenced by the initial condition. Velocity is calculated by tracking the position of the point on the reaction temperature profile where nondimensional temperature is $1/(2\beta)$, which corresponds to 700

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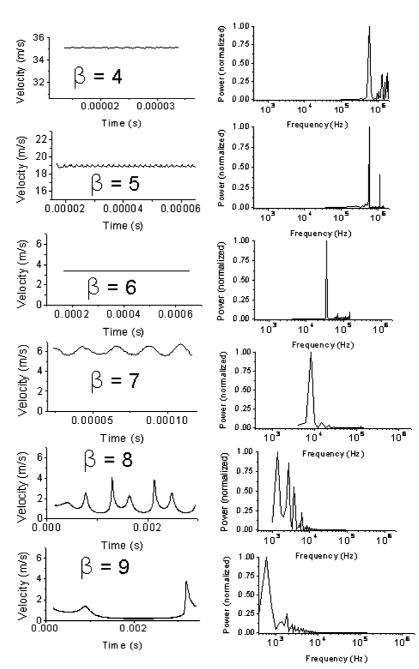


Figure 3. Wavefront velocity profiles and frequency spectra for $4 < \beta < 9$. As β increases, so does the amplitude of the oscillations, and for $\beta > 7$, they begin to take on additional structure and complexity. Stronger secondary peaks in the frequency spectra arise at the same point.

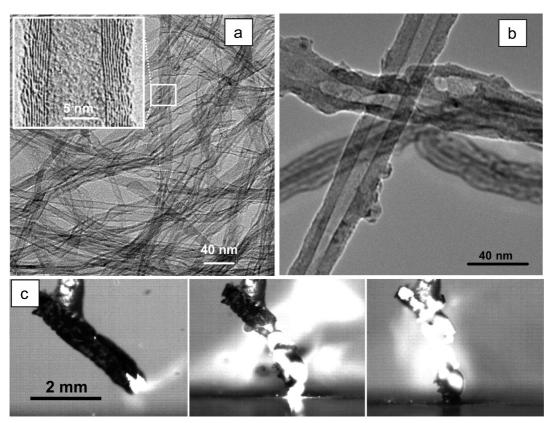
K. The average velocity is 0.56 m/s, although the amplitude of the oscillations is as large as 4.7 m/s. The oscillations are not simply sinusoidal but rather exhibit multiple components at different phase shifts. The mode velocity measured in our previous thermopower experiments was approximately 0.2 m/s,²⁰ so the simulation agrees reasonably well with our measurements. The overprediction is likely due to the adiabatic conditions of the simulation, particularly the exclusion of radiation, which will be further explored in future work.

Including a nanoscale thermal conduit increases the average velocity. The simulated velocity for TNA reacting without CNT thermal conduits is shown in Figure 2b. With no conduit, velocity is on average 0.031 m/s, and the amplitude of oscillations is 0.088 m/s. Including a nanoconduit of $\alpha_2 = 0.001$ m²/s increases the average velocity by a factor of 18 and amplitude by a factor of 53. The rapid thermal transport in the thermal conduits affects the wavefront nonlinearly; they accelerate the reaction wave more when it is moving quickly than when it is moving slowly.

The presence of CNT thermal conduits also decreases the fundamental frequency, as measured using Fourier transforms (FTs) of the velocity profiles. We note that our simulation predicts frequencies on the order of kilohertz. For $\beta = 10.6$, the fundamental frequency

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Figure 4. (a) TEM images of MWNTs and (b) TNA-MWNTs after dispersion. (c, left to right) Reaction front photos using a high-speed camera. The length of the TNA-MWNTs is \sim 3 mm.

decreases from 2350 to 420 Hz after the addition of a thermal conduit, compared to the case without one (Figure 2a,b). This is proportionately smaller than the increase in velocity. It could be that the non-uniform velocity acceleration from the conduits depletes the total energy of the reacting system such that it then takes more time for energy to build up to enable another rapid propagation phase. Future work will develop comprehensive steady-state and transient reactive wave theories that will expand upon these results, but such developments are outside of the scope of this work.

The frequency of these velocity oscillations is inversely related to β , but the dependence is stronger than $1/\beta$, which is what would be expected if it were only due to the time dimensional scaling, $t = (\beta/k_0)\tau$. Figure 2c shows the fundamental frequencies for different β values. As with many properties of the reaction wave, including velocity (see Supporting Information), the relation to β appears to be an inverse exponential of the form $Ae^{-Z\beta}$, where *A* is 6.9×10^7 Hz and *Z* is 1.23 for the best fit to the simulation results. These simulation results indicate that we may be able to control oscillation frequencies in actual devices by using fuels with different properties, corresponding to different β values.

The amplitude of oscillations also depends on β , but not inversely, as frequency and average velocity do, as can be seen in Figure 2d. For $\beta < 7$, velocity os-

cillations are so small as to be inconsequential (although they have well-defined frequencies). Above that point, the relationship is approximately linear; velocity oscillation amplitude increases even as the average velocity is decreasing. This is consistent with the idea that CNT thermal conduits accelerate the reaction wave more when it is moving faster.

The value of β defines two distinct propagation regimes related to oscillations. Figure 3 shows velocity profiles and frequency spectra for $4 < \beta < 9$ at a constant value of $\alpha_2 = 0.001 \text{ m}^2/\text{s}$. Although the average steady-state velocity decreases roughly exponentially from 35 to 0.56 m/s as β increases from 4 to 10.6, the change in oscillations is more complex and transitions from very low amplitude, almost constant velocity profiles to significant oscillations, often with multiple frequency components. This transition occurs between β = 6 and β = 7 and is also marked by the appearance of stronger secondary peaks and harmonics in the freguency spectra. Furthermore, we observe a local minimum in average velocity at $\beta = 6$, which coincides with the transition between propagation regimes. For $\beta >$ 8, the velocity oscillations are interspersed with regions of very low, nearly constant velocity.

Experimental Observations of Thermopower Wave Oscillations. Vertically aligned multiwalled carbon nanotubes (MWNTs) served as the thermal conduits for our thermopower wave experiments. Figure 4a shows transmission electron micrographs (TEM) of the MWNTs after

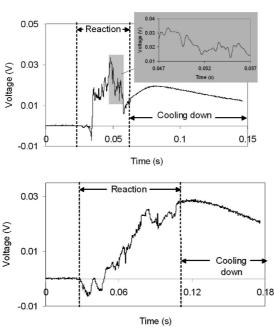


Figure 5. Thermopower voltage signals in reaction (oscillation) and cooling zones. Voltage graphs for (top) a small (0.2 mg) and (bottom) a large mass (3 mg) sample of TNA–MWNTs.

dispersion. They have, on average, an outer diameter of 22 nm, an inner diameter of 14 nm, and 10 walls. The arrays of these MWNTs were 3–5 mm tall with porosity of approximately 99% prior to the addition of TNA. Figure 4b shows the MWNTs after they have been coated with TNA; the continuous annuli are on average 7 nm thick and do not disrupt the MWNTs structurally. After evaporation, the porosity of the arrays was about 90%. Reaction and thermal waves propagated rapidly along the aligned MWNT arrays, as depicted in Figure 4c. Further experimental details can be found in the Methods section.

The accelerated reaction wave drives a concomitant wave of electrons, which also exhibits oscillations in voltage. As an example, Figure 5 shows voltages measured with an oscilloscope across two samples of different masses (0.2 and 3 mg). Large temperature gradients generate voltages with peak magnitudes in the range of 20 to 150 mV. Oscillations modulated this voltage with amplitudes in the range of 2 to 20 mV.

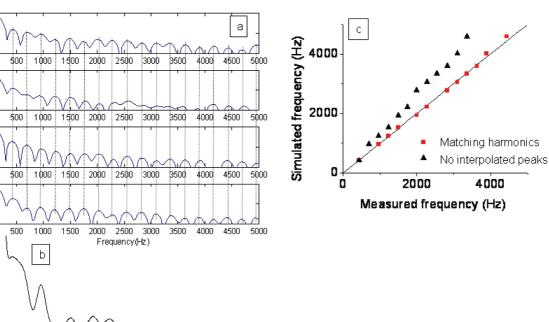
The voltage profiles in Figure 5 can be divided into two phases: a strong initial reaction phase (oscillation/ noise) and a cooling (smooth) phase. Comparison between high-speed video and the electrical signals shows that the oscillation region corresponds to the time the reaction waves are propagating. The shape of the voltage profiles appears to depend on the sample mass: the larger the mass, the longer the oscillation phase. The longer voltage rise time in larger mass samples can be ascribed to slower reaction orthogonal to the MWNTs compared to axial propagation. Despite their very large axial thermal conductivity,³³ MWNT arrays' radial thermal conductivities are very low. In a TNA-MWNT system with TNA layers only a few nanometers thick and nanotube wall spacing of less than 1 nm, as well as large voids between nanotubes, phonon scattering in radial directions is substantial.³⁴ The maximum specific power exceeded 7 kW/kg for smaller mass samples, which is substantially larger than that of highperformance Li-ion batteries.

In agreement with our simulations, the voltage oscillations measured are likewise in the frequency range of 0-5000 Hz. The Fourier transforms of voltage signals from the oscillating reaction phases of four different samples of different masses are shown in Figure 6a. Approximately 40% of all samples had similar frequency signatures, whereas the remainder did not exhibit that pattern. The broad peak around zero frequency extends out to 330 Hz. In addition, there are distinct oscillation peaks at 430, 700, 960, 1230, 1490, ... and 4960 Hz. Except for the first two peaks, the peaks are equally spaced by \sim 260 Hz. These samples came from three different MWNT arrays, had varying cross-sectional areas and masses, and their densities varied up to 20%. However, interestingly, they all show similar frequency signatures. Since the same CVD process produced all samples, the MWNT diameter is consistent between samples (as confirmed by TEM), and thus the (per nanotube) thermal diffusivity is similar between samples.

As can be seen in Figure 6b, the FT of the graph of simulated velocity (Figure 2a) has frequency peaks at very similar positions. Some of these peaks are harmonics that have been broadened and shifted due to noise or finite sampling time, but the velocity oscillations also have multiple fundamentals. The peaks appear at 420, 970, 1250, 1530, ... and 4590 Hz, intervals of ~300 Hz, although not all peaks experimentally observed appear in the simulation.

The parity plot in Figure 6c demonstrates the agreement between the frequencies of the calculated velocity oscillations at $\beta = 10.6$ and the frequencies observed in the experimental voltage oscillations. A simple pairing of fundamental peaks and higher harmonics yields less than a 4% error between the simulated and experimental frequencies, corresponding to the "matching harmonics" data set. There appear to be peaks in the experimental FTs that are not reproduced in the simulated frequencies, for example, at 700, 1750, and 2540 Hz. If one instead assumes that *all* peaks experimentally observed constitute the harmonics, the error is larger at 20-40%, corresponding to the "no interpolated peaks" data set. There is sufficient evidence to support the former analysis (i.e., matching up harmonics). A careful examination of the FT in Figure 6b shows that the 700 Hz harmonic forms the "shoulder" of the 420 Hz peak, and the magnitude of higher frequency harmonics may not be intense enough to detect above the spectrum background. Note that the points in the parity plot are parametric in the numerical order of each peak. With either comparison, however, it is clear that

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Power (arbitrary units) 0 1000 1500 2000 2500 3000 3500 4000 4500 5000 500 Frequency (Hz)

100

50

100

50

100

50

01

100

50

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Power (arbitrary units)

Figure 6. (a) Frequency signatures of reaction region of voltage signals of four samples with different masses (top to bottom: 0.2, 0.9, 1.5, and 3 mg). The dotted lines show the signatures' agreement. Distinct frequency peaks are observed at 430, 700, 960, ... and 4960 Hz. (b) Fourier transform of the simulated velocity. Frequency peaks appear at 420, 970, 1250, ... and 4590 Hz. (c) Parity plot showing agreement between frequency peaks from velocity simulations and voltage measurements. The parameter changing between points is the number of the frequency peak. The agreement improves by matching the harmonics in the two spectra.

there are discretized frequencies predicted by the Fourier analysis and observed experimentally. This supports the accuracy of the mathematical model, which predicts both average velocity (Figure 2a), 0.56 m/s, as well as the frequency of oscillations. Average velocities measured in thermopower experiments ranged from 0.05 to 1.35 m/s, and a range of values of β and α_2 could reproduce these.²⁰ It is notable, therefore, that the results of simulations using $\beta=$ 10.6, $\alpha_2=$ 0.001 m²/s, and $k_0 = 3 \times 10^9 \, \text{s}^{-1}$ agreed well with measurements of velocity, as well as with the frequency spectrum of voltage oscillations.

The parity between the frequencies of the velocity oscillations and the induced voltage implies that the wave velocity is the dominant factor in driving the current along the device. In the limit of conventional, steady-state thermopower (where wave velocity has no influence), the produced voltage, V, is based on the temperature difference alone:

$$V = \kappa (T_{\rm rxn} - T_{\rm ambient}) \tag{7}$$

Here κ is the Seebeck coefficient and \mathcal{T}_{rxn} is the reaction temperature. Note that this expression possesses a Fourier transform that necessarily integrates to zero because T_{rxn} is approximately constant. The fact that V instead has a nonzero Fourier transform, related to the oscillations of the physical wavefront, supports the existence of a velocity component to the generated thermopower. This concept of a velocity component is con-

sistent with our previous observations that samples producing larger velocity (evident in smaller mass samples, for example) generate higher specific power. The origin of the velocity component is currently under investigation, but one physical picture that is suggestive is that of a carrier entrainment that accompanies the wave propagation. Electron-phonon coupling in nanoscale systems is known to be enhanced, with carbon nanotubes being no exception.³⁵ One can calculate an order-of-magnitude estimate of such an entrainment current, J:

$$J = \frac{V}{R_{\rm e}} = n \times A \times v \times e \tag{8}$$

where R_{e} is the resistance of the bundle, *n* is the carrier density of one CNT, A is the cross-sectional area of the bundle, and e is the elementary charge, and v remains the thermopower wave velocity. For a typical current observed from systems experimentally tested to date (0.1-1 mA), we calculate a linear carrier density, nA, of 0.6 to 6 \times 10¹⁶ carriers/m or 0.6 to 6 carriers per nanometer of nanotube since there are about 10⁷ nanotubes per sample. Given the size of a typical CNT unit cell (<1nm), this estimate is reasonable.

In summary, we have both experimentally and theoretically investigated the behavior of the thermopower wave microgenerator. MWNT conduits axially accelerate the highly exothermic decomposition of TNA, generating voltage and current. The maximum DC specific

power of the thermopower wave system exceeds 7 kW/kg, which is larger than any generated by current Liion batteries. The system also shows strong autooscillatory behavior, described by chemical reaction and heat transfer equations and suitable boundary conditions, and these equations can accurately reproduce measured frequency signatures in the range of 0.4 to 5 kHz. This system represents a new class of microscale power sources for applications in which very large power densities are required, within limited space.

METHODS

As with our previous simulations,²⁰ we solved eqs 1, 2, and 3 using the COMSOL Multiphysics software package with adaptive time steps and at various spatial mesh sizes (8–250 nm) over lengths of 2–5 mm to check for convergence. The diameter of the nanotube was 22 nm, and the surrounding fuel annulus was 7 nm thick, matching the physical system used in this work. The amplitude of the initial pulse could be varied by at least 50% without affecting the solution when a nanotube was present. As examples, the dimensional forms of two initial temperature conditions are calculated here. For $\beta = 4$, $g = 11\,000$ K and w = 8×10^{-11} m². For $\beta = 10.6$, $g = 124\,000$ K and $w = 2.1 \times 10^{-10}$ m². These pulses have a full width at half-maximum of 15 and 24 µm, respectively.

Previous modeling work demonstrated that interfacial heat transfer does not limit the wavefront propagation for $G_0 > 10^5$ W/m²/K, corresponding to γ_1 and $\gamma_2 > 10^{-3}$ as described before.²⁰ The simulations were carried out for durations sufficient to observe steady waves with multiple periods of oscillation. A Matlab program extracted the temperature data and calculated velocity. Our previous modeling work has shown that convection does not significantly affect velocity in the TNA–CNT system, and that averaging property values such as thermal conductivity, enthalpy of reaction, and specific heat over an appropriate temperature range results in agreement with simulation using temperature-dependent parameters.²⁰

For our experiments, we synthesized vertically aligned MWNTs using chemical vapor deposition in a horizontal quartz tube furnace with Fe and AI_2O_3 catalytic layers. Ethylene (C_2H_4), hydrogen (H₂), and argon (Ar) were used as the carbon source, catalytic, and carrier gases, respectively.36 To coat the MWNT arrays, a quantity of 200 mg of TNA was dissolved in 10 mL of acetonitrile and filtered. This solution was dropped on the arrays and then evaporated under atmospheric conditions for several hours, coating the nanotubes. NaN₃ in aqueous solution (50 mg/mL) was then added to serve as a primary igniter to lower activation energy (40 kJ/mol for NaN₃, in contrast to 130-200 kJ/ mol for TNA). Silver paste was used to make electrical connections between the TNA-MWNT samples and copper tape electrodes. Their resistance ranged from 10 to 100 Ω depending on the size of the TNA-MWNT bundle. Over 100 samples of different masses were prepared and tested as reported previously.²⁰

The samples were ignited at one end of the bundles with a 400 mW laser of 785 nm wavelength. The propagation of the reaction front was measured optically using a high-speed CCD camera (CPL-MS70K, Canadian Photonic Laboratories) with a microscopic lens (Macro 60 mm, f/2.8D micro Nikkor Autofocus lens, Nikon) at up to 90 000 frames per second.

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Supporting Information Available: Further description and details of the numerical solution methods, as well as a summary of the relation between average velocity and β . This material is available free of charge via the Internet at http://pubs.acs.org.

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